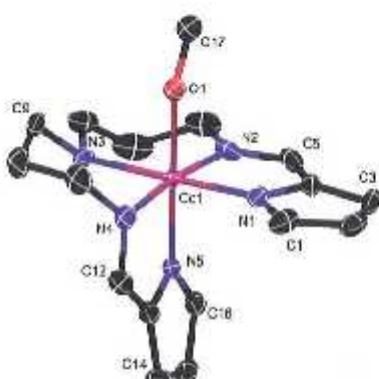


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VOLUME 55A**NUMBER 2****FEBRUARY 2016****CONTENTS**

- 137** Synthesis, crystal structure, redox property and theoretical studies of a pyrrole containing cobalt(III) Schiff base compound

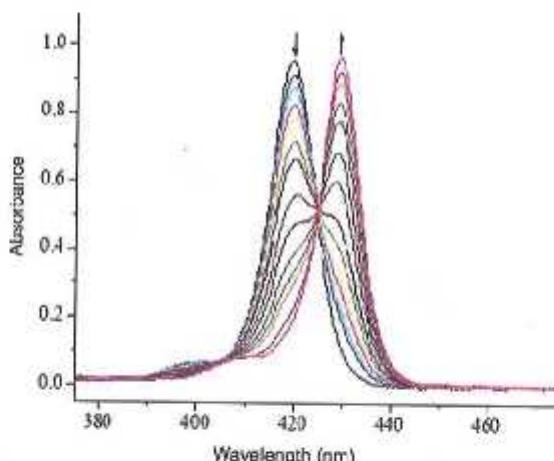
A mononuclear cobalt(III) complex with a pentadentate Schiff base ligand derived from the condensation of N-(3-aminopropyl)-propane-1,3-diamine with pyrrole-2-aldehyde has been synthesised. X-ray crystallography reveals that the geometry of cobalt(III) ion is a distorted octahedron in which the triamine part of the pentadentate ligand occupies the meridional positions, while two pyrrole nitrogen atoms are in *cis* dispositions.



Ananganobhan Panja* & Tarun Kanti Mandal

- 145** Synthesis of zinc porphyrins and effect of peripheral substituents on the coordination reaction

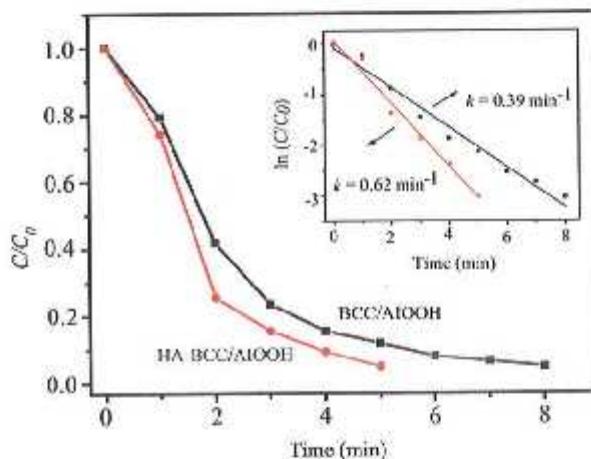
Free-base porphyrins modified with Boc- ω -threonine and their zinc porphyrins have been synthesized. The reaction is spontaneous, with coordinated bonds formed between zinc porphyrins and imidazole derivatives. The ability to form complexes is related to the peripheral substituents in zinc porphyrins and the structure of imidazole derivatives. The zinc porphyrin synthesized with 4-chlorobenzaldehyde and N-methylimidazole shows the highest association constant.



Shujun Wang*, Yuling Peng, Chenggen Zhang,
 Yongbing Li & Chao Liu

- 153 Humic acid-induced synthesis of hierarchical basic copper carbonate/AlOOH microspheres and its enhanced catalytic activity for 4-nitrophenol reduction

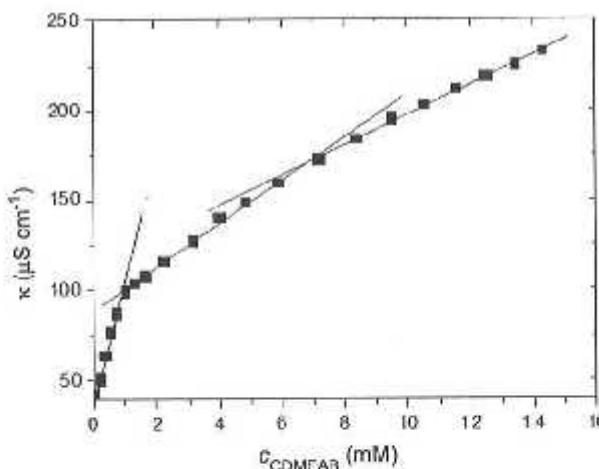
The urchin-like HA-BCC/AlOOH (0.62 min^{-1}) shows better performance than pure BCC/AlOOH microspheres with flower-like structure (0.39 min^{-1}) for reduction of 4-nitrophenol to 4-aminophenol with excess NaBH₄ as a model reaction. Also, addition of humic acid during synthesis of the catalysts greatly enhances the catalytic efficiency.



Wenjin Zhang, Zhengbin Tian, Lijian Chen & Shiyun Ai*

- 160 Micellar parameters and thermodynamics of interaction of fluoroquinolone drugs with cetyltrimethylammonium bromide

Interaction of fluoroquinolone antibiotic drugs, viz., ciprofloxacin hydrochloride, levofloxacin hemihydrate and lomefloxacin hydrochloride, with the cationic surfactant, cetyltrimethylammonium bromide, is studied by conductance measurements in water and in the presence of salts such as NaCl, Na₂SO₄ and Na₃PO₄.12H₂O over the temperature range of 298.15–318.15 K. Addition of drug alters the micellization behaviour of the surfactant. Favourable micellization is observed in the presence of salts. Thermodynamic parameters reveal that drug-CDMEAB interactions are mainly hydrophobic and electrostatic in nature.

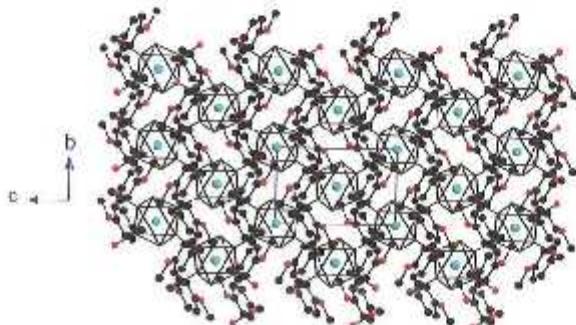


Sk. Md. Ali Ahsan, Mohammed Delwar Hossain, Md. Anamul Hoque* & Mohammed Abdullah Khan

Notes

- 170** Tetraaquabis(4-methoxyphenylacetato-O')magnesium(II) dihydrate and catena-poly[[diaqua)manganese(III)]-bis(μ_2 -4-methoxyphenylacetato-O,O'): A monomeric and a two-dimensional coordination polymer based on 4-methoxyphenylacetic acid

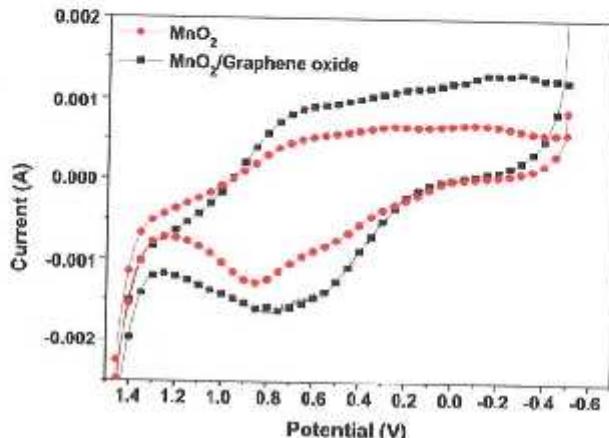
In tetraaquabis(4-methoxyphenylacetato-O')magnesium(II) dihydrate, the 4-mpa functions as a monodentate ligand for Mg(II) situated on an inversion centre, resulting in a monomeric Mg(II) complex. The Mn(II) in catena-poly[[diaqua)-manganese(II)]-bis(μ_2 -4-methoxyphenylacetato-O,O')] is also located on an inversion centre and the μ_2 -bridging bidentate coordination mode of the ligand results in formation of a two-dimensional Mn(II) coordination polymer.



Kiran T Dhavskar & Bikshandarkoil R Srinivasan*

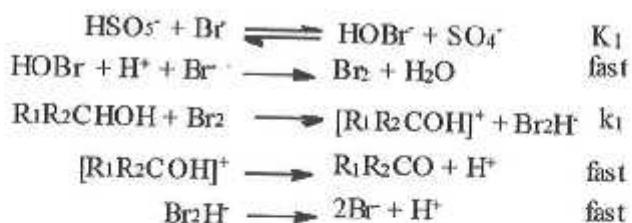
- 177** Oxygen reduction reaction of manganese oxide/graphene oxide nanocomposite

MnO₂/graphene oxide composite coating has been prepared by electrochemical anodic deposition method. Morphological studies reveal the increase in porosity and exfoliation of graphene oxide by MnO₂ particles. The MnO₂/GO composite coating has low over-potential for oxygen reduction as compared to MnO₂.



K Pravinkumar, S Balaji*, T Manichandran,
M Anandakumar & G Kumaraguruparan

- 182** Kinetics and mechanism of oxidation of aliphatic and aromatic alcohols by *in situ* generated bromine in reaction between ozone and bromide ion



R₁ = H for primary aliphatic alcohols

C₆H₅ for aromatic alcohols

Malharrao R Thombare & Gavisiddappa S Gokavi*