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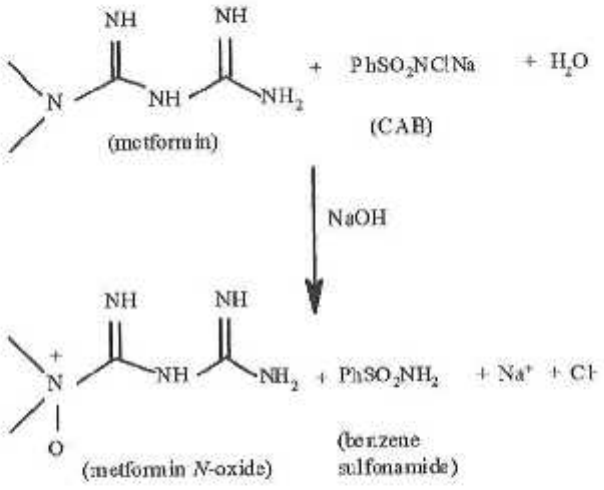
CONTENTS

Notes

484 **Oxidation of metformin with alkaline chloramine-B: Delineation of reaction mechanism and kinetic modeling**

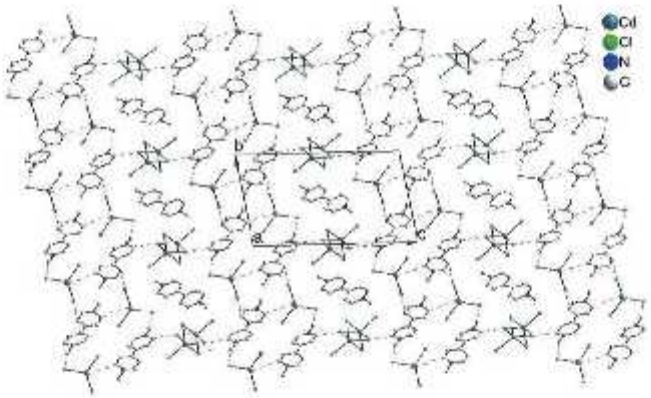
The kinetics and mechanism of oxidation of metformin hydrochloride with chloramine B have been investigated in alkaline medium at 308 K. The reaction rate shows a first order dependence of rate each on $[CAB]_o$ and $[MET]_o$, and an inverse-first order dependence on $[OH^-]$. The conjugate acid ($PhSO_2NHCl$) is postulated as the reactive oxidizing species.



Nirmala Vaz*, A S Manjunath & Puttaswamy

489 **Photoluminescence and theoretical study of $[^{\infty}_{1}(Cd_2Cl_6)][2(CdCl_4)][3(N,N'\text{-dimethyl-}4,4'\text{-bipyridinium})]$**

A novel cadmium chloride-4,4'-bipyridinium compound, namely $[^{\infty}_{1}(Cd_2Cl_6)][2(CdCl_4)][3(N,N'\text{-dimethyl-}4,4'\text{-bipyridinium})]$ in which the N,N'-dimethyl-4,4'-bipyridinium has been prepared *in situ*, has been obtained through solvothermal reaction. The compound contains a one-dimensional infinite $^{\infty}_{1}(Cd_2Cl_6)$ chain, $CdCl_4^{2-}$ anions and $[N,N'\text{-dimethyl-}4,4'\text{-bipyridinium}]^{2+}$ cations, interlinked via hydrogen bonding interactions to yield a three-dimensional supramolecular framework.



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